2'(3')-O-[N- [2- [3- [5-fluoresceinyl] thioureido] ethyl] carbamoyl] adenosine 5'-triphosphate and its $Cr(H_2O)_4$ and $Co(NH_3)_4$ Complex Derivatives are New Fluorescent Tools for Labelling ATP Binding Sites of Na⁺/K⁺-ATPase

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Received March 5, 1997 Accepted May 14, 1997

Summary

2'(3')-O-[N- [2- [3- [5-fluoresceinyl] thioureido] ethyl] carbamoyl] adenosine 5'-triphosphate (FEDA-ATP), a spectroscopic tool used for studying skeletal muscle myosin ATPase subfragment 1, was applied to Na⁺/K⁺-ATPase (EC 3.6.1.37). In contrast to the myosin subfragment, we found that FEDA-ATP is not a substrate for Na⁺/K⁺-ATPase. On the other hand, FEDA-ATP showed an affinity for both the low (E₂, K_d=200 μ M) and the high (E₁, K_d=22 μ M) affinity ATP-binding sites. When the microscopic affinities of FEDA-ATP were used for calculating the macroscopic affinity in the overall reaction according to K_i = (K_{dE1}*K_{dE2})^{1/2}, the experimentally measured inhibition constant of 66 μ M was obtained. To evoke irreversible binding inhibitors, FEDA-ATP was transferred in its chromium(III) and cobalt(III) complex analogs, which are suitable tools for labelling the ATP binding sites of Na⁺/K⁺-ATPase in a specific way.

Key words

 Na^+/K^+ -ATPase – Complex derivatives of ATP – Low-affinity ATP-binding site – High-affinity ATP binding site

Abbreviations

 $Cr(H_2O)_4ATP, \beta, \gamma$ bidentate complex of chromium (III)-tetraaqua-adenosine-5'-triphosphate

Cr(H₂O)₄FEDA-ATP, β, γ bidentate complex of chromium (III)-tetraaqua- 2'(3')-O-[N- [2- [3- [5- fluoresceinyl] thioureido] ethyl] carbamoyl] adenosine-5'-triphosphate

 $Co(NH_3)_4ATP, \beta, \gamma$ bidentate complex of cobalt (III)-tetramino-adenosine-5'-triphosphate

 $Co(NH_3)_4FEDA-ATP, \beta, \gamma$ bidentate complex of cobalt (III)-tetramino- 2'(3')-O-[N- [2- [3- [5- fluoresceinyl] thioureido] ethyl] carbamoyl]- adenosine-5'-triphosphate

Co(NH₃)₄PO₄, tetramine cobalt(III)phosphate

FITC, fluorescein 5'-isothiocyanate

FEDA-ATP, 2'(3')-O-[N- [2- [3- [5- fluoresceinyl] thioureido] ethyl] carbamoyl] adenosine 5'-triphosphate EDA-ATP, 2'(3')-O-[amino ethyl carbamoyl] adenosine 5'-triphosphate

 E_1 ATP site, nucleotide-binding site of Na⁺/K⁺-ATPase with high affinity for ATP

E₂ATP site, nucleotide-binding site of Na⁺/K⁺-ATPase with low affinity for ATP

Introduction

Na⁺/K⁺-ATPase (EC 3.6.1.37) is the biochemical equivalent of the sodium pump of mammalian plasma membranes (Skou and Esmann 1992). The enzyme probably works as a functional $(\alpha\beta)_2$ -diprotomer (Schoner *et al.* 1994). Kinetic studies on ATP hydrolysis revealed that phosphorylation of the catalytic α -subunit at Asp-369 occurs in the presence of Na⁺ from a high affinity ATP site ($K_d = 1 \mu M$, $E_1 ATP$ site). This process is followed by K⁺-activated of the phospho-intermediate. Na⁺hydrolysis dependent phosphorylation of the α -subunit leads to the occlusion of three Na⁺ ions. These ions are released into the external medium during the reaction cycle. Occlusion of two K⁺ ions, however, takes place in the nonphosphorylated α subunit. To release K⁺ at the inner membrane face, ATP has to bind to a lowaffinity ATP site (E₂ATP site), which hydrolyzes p-nitrophenylphosphate and supports the binding of ouabain in the presence of P_i (Linnertz et al. 1994). Studies with chromium(III) and cobalt(III) complex analogs of ATP revealed that the E_1ATP and E_2ATP sites coexist and that these MgATP complex analogs can be used to differentiate between them. Cr(H₂O)₄ATP is an almost irreversible specific inhibitor of the E1ATP and Co(NH3)4ATP of the E₂ATP-binding site (Schoner et al. 1994, Linnertz et al. 1995, Linnertz and Schoner 1996).

2'(3')-O-[N- [2- [3- [5-fluoresceinyl] thioureido] ethyl] carbamoyl] adenosine 5'-triphosphate (FEDA-ATP) known from studies of skeletal muscle myosin subfragment 1 (S1) (Conibar *et al.* 1996) makes it possible to determine the activity of single myosin filaments. FEDA-ATP as substrate is hydrolyzed to P_i and FEDA-ADP, which is released with the same rate constant as ADP. As a result of fluorescence changes, it was possible to follow turnover of this ATP analog by fluorescence microscopy (Conibar *et al.* 1996).

The aim of the present work was to ascertain whether FEDA-ATP is a substrate of Na⁺/K⁺-ATPase and whether it and its MgATP complex analogs provide information about the E₁ATP and/or E₂ATP sites. We therefore studied the interaction of FEDA-ATP with the high-affinity E₁ATP site, which is inactivated by Cr(H₂O)₄ATP, and with the low-affinity E₂ATP site, which is inactivated by Co(NH₃)₄ATP. Furthermore, FEDA-ATP was transferred in its chromium(III) and cobalt (III) complex analogs, which are suitable tools for labelling the ATP binding sites of Na⁺/K⁺-ATPase in a specific way.

Materials and Methods

All chemicals were of highest available purity and were obtained from Bio-Rad (Munich, Germany), Boehringer-Mannheim (Mannheim, Germany), E. Merck (Darmstadt, Germany) and Molecular Probes (Eugene, USA). The Lab-Trol protein standard was a product of Merz & Dade (Munich, Germany). $[\gamma^{32}P]ATP$ was from Amersham Buchler (Braunschweig, Germany).

Synthesis of FEDA-ATP, MgATP- and MgFEDA-ATP- complex analogs

Ethylenediamine was first coupled to ATP according to Cremo *et al.* (1990) to yield 2'(3')-O-[amino ethyl carbamoyl] adenosine 5'-triphosphate (EDA-ATP). This intermediate product was then made to react with fluorescein 5'-isothiocyanate (isomer 1) according to Sowerby *et al.* (1993) to yield 2'(3')-O-[N- [2- [3- [5- fluoresceinyl] thioureido] ethyl] carbamoyl] adenosine 5'-triphosphate (FEDA-ATP). The concentration of FEDA-ATP was determined assuming an absorbance coefficient at pH 9.0, A495 of 75 000 M⁻¹ cm⁻¹. Cr(H₂O)₄ATP, Co(NH₃)₄FEDA-ATP and Cr(H₂O)₄FEDA-ATP complex analogs were prepared by the aniline procedure of Cleland and coworkers (De Pamphilis and Cleland 1973, Cornelius *et al.* 1977).

Isolation of Na^+/K^+ -ATPase

 Na^+/K^+ -ATPase was isolated from pig kidney according to the method of Jørgensen (1974) and assayed as described earlier (Schoner *et al.* 1994, Linnertz *et al.* 1995).

Inhibition of ATP hydrolysis by FEDA-ATP

Various concentrations of FEDA-ATP (0–100 μ M) were used in 1 ml of the optical assay mixture (Schoner *et al.* 1967) and ATP concentration range from 25 to 500 μ M.

Protective action of FEDA-ATP against the inactivation by MgATP analogs

Affinities of FEDA-ATP for the E₁ATP and E₂ATP binding sites were determined from the protective effect of FEDA-ATP as a function of its concentration $(0-500 \,\mu\text{M})$. Na⁺/K⁺-ATPase (0.5 units, 20 units/mg protein) was incubated in a total volume of 250 μ l at 37 °C, 15 mM Tris/HCl buffer (pH 7.5) and various concentrations of Cr(H₂O)₄ATP $(0-100 \,\mu\text{M})$ or Co(NH₃)₄ATP $(0-500 \,\mu\text{M})$. Residual Na⁺/K⁺-ATPase activity was measured at the time indicated by transferring aliquots of 30 μ l to the optical assay (Schoner *et al.* 1967).

Inactivation with $Cr(H_2O)_4FEDA$ -ATP and $Co(NH_3)_4FEDA$ -ATP

One unit of Na⁺/K⁺-ATPase was incubated in a total volume of 1 ml at 37 °C with 20 mM Tris/HCl buffer at pH 7.25, 15 mM NaCl and different concentrations of the inhibitor. Residual activity was measured by transferring 50 μ l aliquots to the optical assay mixture (Schoner *et al.* 1967).



Fig. 1. Inhibition of ATP hydrolysis by FEDA-ATP. (A) ATP hydrolysis was measured in 1 ml of the optical assay mixture (Schoner et al. 1967) containing $25-500 \mu M$ ATP and various concentrations of FEDA-ATP (full square $0 \mu M$, open square $20 \mu M$, full circle $40 \mu M$, open circle $60 \mu M$ and full rhombus $100 \mu M$). We detected a reduction of the maximum velocity of ATP-hydrolysis. (B) The inhibition of ATP-hydrolysis by FEDA-ATP results in an inhibition constant of $66 \mu M$ and a maximum velocity of 26.7 U/ml.

Results

Pig kidney Na⁺/K⁺-ATPase (1 U/ml) was incubated with 0, 20, 40, 60 and $100 \,\mu\text{M}$ FEDA-ATP and rates of ATP hydrolysis were measured by the optical assay. Hydrolysis of ATP was inhibited by FEDA-ATP (Fig. 1A), at 100 µM FEDA-ATP, being about 25 % of the control. The inhibition constant K_i was found to be $66 \,\mu M$ (Fig. 1B). However, this macroscopic affinity of FEDA-ATP of 66 µM in the overall reaction is the result of the microscopic affinities of the substance for the E_1ATP and E_2ATP binding sites (Thoenges et al. 1997). To establish whether FEDA-ATP binds preferentially to a specific ATP site, its protective effects against inactivation with the specific inhibitor of the E₁-site, namely $Cr(H_2O)_4ATP$, was studied. Na⁺/K⁺-ATPase

(1 U/ml) was incubated in a mixture of 20 mM Tris/HCl (pH 7.25) and 15 mM NaCl at 37 °C with $0\,\mu$ M, 50 mM and 100 mM Cr(H₂O)₄ATP and the enzyme inhibition curves were measured (Fig. 2).



Fig. 2. Protective effect of different FEDA-ATP concentrations against the inactivation by $Cr(H_2O)_4ATP$. One unit Na^+/K^+ -ATPase was incubated in a total volume of 250 µl with different concentrations of $Cr(H_2O)_4ATP$ (0, 20, 40, 60, 80 and 100 µM) and FEDA-ATP (full square 0 µM, open circle 50 µM and full rhombus 100 µM). k_{app} are the apparent velocity constants.



Fig. 3. Protective effect of different FEDA-ATP concentrations against the inactivation by $Co(NH_3)_4ATP$. One unit Na^+/K^+ -ATPase was incubated in a total volume of 250 µl with different concentrations of $Co(NH_3)_4ATP$ (0, 320, 480, 640, 800 and 960 µM) and FEDA-ATP (full square 0 µM, open circle 250 µM and full rhombus 500 µM). k_{app} are the apparent velocity constants.

Apparently, FEDA-ATP exhibited a competitive and protective effect against the enzyme inactivation by $Cr(H_2O)_4ATP$. The dissociation constant K_d for the high-affinity ATP-binding site (E₁-site) was calculated as $22 \,\mu$ M.

A similar experiment was performed for the low-affinity ATP-binding site (E₂-site). In this case, the Na⁺/K⁺-ATPase inhibition curves were determined with increasing concentrations of Co(NH₃)₄ATP in the absence and in the presence of $250 \,\mu$ M and $500 \,\mu$ M FEDA-ATP. Similarly to Cr(H₂O)₄ATP, a competitive and protective effect of FEDA-ATP against Co(NH₃)₄ATP was observed (Fig. 3). The calculated dissociation constant in this case (low affinity site), however, was by about one order of magnitude higher (K_d=0.2 mM).

The extinction coefficient of FEDA-ATP in 20 mM Tris/HCl buffer (pH 9) was $A_{490}=78,000 \text{ M}^{-1}$ cm⁻¹. The fluorescence measurements in 20 mM Tris/HCl buffer (pH 7.25) result in an $\lambda_{ex}=495$ nm and $\lambda_{em}=520$ nm. We titrated a solution of 1 μ M FEDA-ATP with increasing concentrations of Na⁺/K⁺-ATPase but were not able to detect any significant change of the steady-state fluorescence intensity even with a 10-fold excess of the enzyme (data not shown).



Fig. 4. Inactivation of Na^+/K^+ -ATPase activity by $Cr(H_2O)_4FEDA$ -ATP. One unit of Na^+/K^+ -ATPase was incubated in a total volume of 1 ml at 37 °C with 20 mM Tris/HCl-buffer pH 7.25, 15 mM NaCl and various concentrations of the inhibitor (full square control, open square 0.3 μ M, full circle 1.3 μ M and open circle 5.0 μ M $Cr(H_2O)_4FEDA$ -ATP). Residual activity was measured by transferring aliquots of 50 μ l to the optical assay mixture (Schoner et al. 1967).

To bring about irreversible binding, we prepared analogs of FEDA-ATP with chromium(III) and cobalt (III) and studied their inactivation effect on Na⁺/K⁺-ATPase. One unit of the enzyme was incubated in 20 mM Tris/HCl (pH 7.25), 15 mM NaCl

at 37 °C with increasing concentrations of Cr(H₂O)₄FEDA-ATP (Fig. 4) and Co(NH₃)₄FEDA-ATP (Fig. 5). Aliquots of 50 μ l were taken from the incubation medium for the residual Na⁺/K⁺-ATPase activity measurement by the optical assay. The Mg²⁺ complex derivatives of FEDA-ATP behaved similarly to those of ATP: 5 μ M Cr(H₂O)₄FEDA-ATP inactivated by 55 % in 3 h (Fig. 4), while 6 μ M Co(NH₃)₄FEDA-ATP inactivated by 30 % only (Fig. 5).



Fig. 5. Inactivation of Na^+/K^+ -ATPase activity by $Co(NH_3)_4FEDA$ -ATP. One unit of Na^+/K^+ -ATPase was incubated in a total volume of 1 ml at 37 °C with 20 mM Tris/HCl-buffer pH 7.25, 15 mM NaCl and various concentrations of inhibitor (full square control, cross 3.0 μ M, full rhombus 6.0 μ M and open rhombus 12.5 μ M $Co(NH_3)_4FEDA$ -ATP). Residual activity was measured by transferring aliquots of 50 μ l to the optical assay mixture (Schoner et al. 1967).

Discussion

FEDA-ATP does not appear to be a substrate of Na⁺/K⁺-ATPase isolated from the pig kidney but rather a reversible competitive inhibitor of ATP hydrolysis. The presence of FEDA-ATP resulted in enzyme inhibition but the steady-state fluorescence intensity was not changed significantly. This is in contrast with previous reports on myosin fragment S1 (Conibar *et al.* 1996) where FEDA-ATP was reported as a substrate with changed fluorescence properties after binding. This could indicate differences in the microenvironments of the binding sites of the two systems. Nevertheless, additional steady-state and dynamic fluorescence studies will have to be performed.

On the other hand, FEDA-ATP showed the affinity for both the low (E₂, K_d=200 μ M) and the high (E₁, K_d=2 μ M) affinity ATP-binding sites. Apparently, the FEDA-ATP behaves like ATP in these reactions (Schoner *et al.* 1994). When the microscopic affinities of FEDA-ATP were used to calculate the macroscopic

affinity in the overall reaction according to $K_i = (K_{dE1} * K_{dE2})^{1/2}$ given by Thoenges *et al.* (1997), the experimentally measured inhibition constant of 66 μ M was obtained.

In addition, the chromium(III) and cobalt(III) analogs of FEDA-ATP behaved similarly as the analogs of ATP. The chromium analog of FEDA-ATP inhibited the high affinity ATP-binding site while the cobalt analog of FEDA-ATP was found to inhibit the low-affinity ATP-binding site. These observations open up the possibility for labelling specifically the E₁ATP site with $Cr(H_2O)_4$ FEDA-ATP and to introduce another label on the E₂ATP site as a Co(NH₃)₄ATP derivative. Since 2'(3')-O-[amino ethyl carbamoyl] adenosine 5'-triphosphate (EDA-ATP) can react with any other isothiocyanate but FITC (Cremo *et al.* 1990), other fluorescent ATP analogs overlapping in their fluorescent spectra with fluorescein are conceivable as $Co(NH_3)_{4}$ complex derivatives. Such analogs may help to determine the distance between the E_1ATP and E_2ATP sites by Förster energy transfer.

Acknowledgements

The authors thank Mr. Wolfgang Mertens for technical assistance and to Prof. Arnošt Kotyk for valuable comments for improving this manuscript. This work is part of the Ph.D. thesis of Holger Linnertz at the Justus-Liebig-University, Giessen. The work was supported by the DFG (Bonn) through Graduiertenkolleg "Molekulare Biologie und Pharmakologie" Giessen, the Fonds der Chemischen Industrie (Frankfurt/M), the German government through "bilaterale wissenschaftliche Zusammenarbeit" and by the grants of GA ČR No. 204/95/0624 and of GA AV ČR A511403.

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