2'(3')-O-[N- [2- [3- [5-fluoresceinyl] thioureido] ethyl] carbamoyl] adenosine 5'-triphosphate and its $Cr(H_2O)_4$ and $Co(NH_3)_4$ Complex Derivatives are New Fluorescent Tools for Labelling ATP Binding Sites of Na⁺/K⁺-ATPase

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Summary

2'(3')-O-[N- [2- [3- [5-fluoresceinyl] thioureido] ethyl] carbamoyl] adenosine 5'-triphosphate (FEDA-ATP), a spectroscopic tool used for studying skeletal muscle myosin ATPase subfragment 1, was applied to Na⁺/K⁺-ATPase (EC 3.6.1.37). In contrast to the myosin subfragment, we found that FEDA-ATP is not a substrate for Na⁺/K⁺-ATPase. On the other hand, FEDA-ATP showed an affinity for both the low (E₂, K_d=200 μ M) and the high (E₁, K_d=22 μ M) affinity ATP-binding sites. When the microscopic affinities of FEDA-ATP were used for calculating the macroscopic affinity in the overall reaction according to K_i = (K_{dE1}*K_{dE2})^{1/2}, the experimentally measured inhibition constant of 66 μ M was obtained. To evoke irreversible binding inhibitors, FEDA-ATP was transferred in its chromium(III) and cobalt(III) complex analogs, which are suitable tools for labelling the ATP binding sites of Na⁺/K⁺-ATPase in a specific way.

Key words

 Na^+/K^+ -ATPase – Complex derivatives of ATP – Low-affinity ATP-binding site – High-affinity ATP binding site

Abbreviations

 $Cr(H_2O)_4ATP, \beta, \gamma$ bidentate complex of chromium (III)-tetraaqua-adenosine-5'-triphosphate

Cr(H₂O)₄FEDA-ATP, β, γ bidentate complex of chromium (III)-tetraaqua- 2'(3')-O-[N- [2- [3- [5- fluoresceinyl] thioureido] ethyl] carbamoyl] adenosine-5'-triphosphate

 $Co(NH_3)_4ATP, \beta, \gamma$ bidentate complex of cobalt (III)-tetramino-adenosine-5'-triphosphate

 $Co(NH_3)_4FEDA-ATP, \beta, \gamma$ bidentate complex of cobalt (III)-tetramino- 2'(3')-O-[N- [2- [3- [5- fluoresceinyl] thioureido] ethyl] carbamoyl]- adenosine-5'-triphosphate

Co(NH₃)₄PO₄, tetramine cobalt(III)phosphate

FITC, fluorescein 5'-isothiocyanate

FEDA-ATP, 2'(3')-O-[N- [2- [3- [5- fluoresceinyl] thioureido] ethyl] carbamoyl] adenosine 5'-triphosphate EDA-ATP, 2'(3')-O-[amino ethyl carbamoyl] adenosine 5'-triphosphate

 E_1 ATP site, nucleotide-binding site of Na⁺/K⁺-ATPase with high affinity for ATP

E₂ATP site, nucleotide-binding site of Na⁺/K⁺-ATPase with low affinity for ATP

Introduction

Na⁺/K⁺-ATPase (EC 3.6.1.37) is the biochemical equivalent of the sodium pump of mammalian plasma membranes (Skou and Esmann 1992). The enzyme probably works as a functional $(\alpha\beta)_2$ -diprotomer (Schoner *et al.* 1994). Kinetic studies on ATP hydrolysis revealed that phosphorylation of the catalytic α -subunit at Asp-369 occurs in the presence of Na⁺ from a high affinity ATP site ($K_d = 1 \mu M$, $E_1 ATP$ site). This process is followed by K⁺-activated of the phospho-intermediate. Na⁺hydrolysis dependent phosphorylation of the α -subunit leads to the occlusion of three Na⁺ ions. These ions are released into the external medium during the reaction cycle. Occlusion of two K⁺ ions, however, takes place in the nonphosphorylated α subunit. To release K⁺ at the inner membrane face, ATP has to bind to a lowaffinity ATP site (E₂ATP site), which hydrolyzes p-nitrophenylphosphate and supports the binding of ouabain in the presence of P_i (Linnertz et al. 1994). Studies with chromium(III) and cobalt(III) complex analogs of ATP revealed that the E_1ATP and E_2ATP sites coexist and that these MgATP complex analogs can be used to differentiate between them. Cr(H₂O)₄ATP is an almost irreversible specific inhibitor of the E1ATP and Co(NH3)4ATP of the E₂ATP-binding site (Schoner et al. 1994, Linnertz et al. 1995, Linnertz and Schoner 1996).

2'(3')-O-[N- [2- [3- [5-fluoresceinyl] thioureido] ethyl] carbamoyl] adenosine 5'-triphosphate (FEDA-ATP) known from studies of skeletal muscle myosin subfragment 1 (S1) (Conibar *et al.* 1996) makes it possible to determine the activity of single myosin filaments. FEDA-ATP as substrate is hydrolyzed to P_i and FEDA-ADP, which is released with the same rate constant as ADP. As a result of fluorescence changes, it was possible to follow turnover of this ATP analog by fluorescence microscopy (Conibar *et al.* 1996).

The aim of the present work was to ascertain whether FEDA-ATP is a substrate of Na⁺/K⁺-ATPase and whether it and its MgATP complex analogs provide information about the E₁ATP and/or E₂ATP sites. We therefore studied the interaction of FEDA-ATP with the high-affinity E₁ATP site, which is inactivated by Cr(H₂O)₄ATP, and with the low-affinity E₂ATP site, which is inactivated by Co(NH₃)₄ATP. Furthermore, FEDA-ATP was transferred in its chromium(III) and cobalt (III) complex analogs, which are suitable tools for labelling the ATP binding sites of Na⁺/K⁺-ATPase in a specific way.

Materials and Methods

All chemicals were of highest available purity and were obtained from Bio-Rad (Munich, Germany), Boehringer-Mannheim (Mannheim, Germany), E. Merck (Darmstadt, Germany) and Molecular Probes (Eugene, USA). The Lab-Trol protein standard was a product of Merz & Dade (Munich, Germany). $[\gamma^{32}P]ATP$ was from Amersham Buchler (Braunschweig, Germany).

Synthesis of FEDA-ATP, MgATP- and MgFEDA-ATP- complex analogs

Ethylenediamine was first coupled to ATP according to Cremo *et al.* (1990) to yield 2'(3')-O-[amino ethyl carbamoyl] adenosine 5'-triphosphate (EDA-ATP). This intermediate product was then made to react with fluorescein 5'-isothiocyanate (isomer 1) according to Sowerby *et al.* (1993) to yield 2'(3')-O-[N- [2- [3- [5- fluoresceinyl] thioureido] ethyl] carbamoyl] adenosine 5'-triphosphate (FEDA-ATP). The concentration of FEDA-ATP was determined assuming an absorbance coefficient at pH 9.0, A495 of 75 000 M⁻¹ cm⁻¹. Cr(H₂O)₄ATP, Co(NH₃)₄FEDA-ATP and Cr(H₂O)₄FEDA-ATP complex analogs were prepared by the aniline procedure of Cleland and coworkers (De Pamphilis and Cleland 1973, Cornelius *et al.* 1977).

Isolation of Na^+/K^+ -ATPase

 Na^+/K^+ -ATPase was isolated from pig kidney according to the method of Jørgensen (1974) and assayed as described earlier (Schoner *et al.* 1994, Linnertz *et al.* 1995).

Inhibition of ATP hydrolysis by FEDA-ATP

Various concentrations of FEDA-ATP (0–100 μ M) were used in 1 ml of the optical assay mixture (Schoner *et al.* 1967) and ATP concentration range from 25 to 500 μ M.

Protective action of FEDA-ATP against the inactivation by MgATP analogs

Affinities of FEDA-ATP for the E₁ATP and E₂ATP binding sites were determined from the protective effect of FEDA-ATP as a function of its concentration $(0-500 \,\mu\text{M})$. Na⁺/K⁺-ATPase (0.5 units, 20 units/mg protein) was incubated in a total volume of 250 μ l at 37 °C, 15 mM Tris/HCl buffer (pH 7.5) and various concentrations of Cr(H₂O)₄ATP $(0-100 \,\mu\text{M})$ or Co(NH₃)₄ATP $(0-500 \,\mu\text{M})$. Residual Na⁺/K⁺-ATPase activity was measured at the time indicated by transferring aliquots of 30 μ l to the optical assay (Schoner *et al.* 1967).

Inactivation with $Cr(H_2O)_4FEDA$ -ATP and $Co(NH_3)_4FEDA$ -ATP

One unit of Na⁺/K⁺-ATPase was incubated in a total volume of 1 ml at 37 °C with 20 mM Tris/HCl buffer at pH 7.25, 15 mM NaCl and different concentrations of the inhibitor. Residual activity was measured by transferring 50 μ l aliquots to the optical assay mixture (Schoner *et al.* 1967).



Fig. 1. Inhibition of ATP hydrolysis by FEDA-ATP. (A) ATP hydrolysis was measured in 1 ml of the optical assay mixture (Schoner et al. 1967) containing $25-500 \mu M$ ATP and various concentrations of FEDA-ATP (full square $0 \mu M$, open square $20 \mu M$, full circle $40 \mu M$, open circle $60 \mu M$ and full rhombus $100 \mu M$). We detected a reduction of the maximum velocity of ATP-hydrolysis. (B) The inhibition of ATP-hydrolysis by FEDA-ATP results in an inhibition constant of $66 \mu M$ and a maximum velocity of 26.7 U/ml.

Results

Pig kidney Na⁺/K⁺-ATPase (1 U/ml) was incubated with 0, 20, 40, 60 and $100 \,\mu\text{M}$ FEDA-ATP and rates of ATP hydrolysis were measured by the optical assay. Hydrolysis of ATP was inhibited by FEDA-ATP (Fig. 1A), at 100 µM FEDA-ATP, being about 25 % of the control. The inhibition constant K_i was found to be $66 \,\mu M$ (Fig. 1B). However, this macroscopic affinity of FEDA-ATP of 66 µM in the overall reaction is the result of the microscopic affinities of the substance for the E_1ATP and E_2ATP binding sites (Thoenges et al. 1997). To establish whether FEDA-ATP binds preferentially to a specific ATP site, its protective effects against inactivation with the specific inhibitor of the E₁-site, namely $Cr(H_2O)_4ATP$, was studied. Na⁺/K⁺-ATPase

(1 U/ml) was incubated in a mixture of 20 mM Tris/HCl (pH 7.25) and 15 mM NaCl at 37 °C with $0\,\mu$ M, 50 mM and 100 mM Cr(H₂O)₄ATP and the enzyme inhibition curves were measured (Fig. 2).



Fig. 2. Protective effect of different FEDA-ATP concentrations against the inactivation by $Cr(H_2O)_4ATP$. One unit Na^+/K^+ -ATPase was incubated in a total volume of 250 µl with different concentrations of $Cr(H_2O)_4ATP$ (0, 20, 40, 60, 80 and 100 µM) and FEDA-ATP (full square 0 µM, open circle 50 µM and full rhombus 100 µM). k_{app} are the apparent velocity constants.



Fig. 3. Protective effect of different FEDA-ATP concentrations against the inactivation by $Co(NH_3)_4ATP$. One unit Na^+/K^+ -ATPase was incubated in a total volume of 250 µl with different concentrations of $Co(NH_3)_4ATP$ (0, 320, 480, 640, 800 and 960 µM) and FEDA-ATP (full square 0 µM, open circle 250 µM and full rhombus 500 µM). k_{app} are the apparent velocity constants.

Apparently, FEDA-ATP exhibited a competitive and protective effect against the enzyme inactivation by $Cr(H_2O)_4ATP$. The dissociation constant K_d for the high-affinity ATP-binding site (E₁-site) was calculated as $22 \,\mu$ M.

A similar experiment was performed for the low-affinity ATP-binding site (E₂-site). In this case, the Na⁺/K⁺-ATPase inhibition curves were determined with increasing concentrations of Co(NH₃)₄ATP in the absence and in the presence of $250 \,\mu$ M and $500 \,\mu$ M FEDA-ATP. Similarly to Cr(H₂O)₄ATP, a competitive and protective effect of FEDA-ATP against Co(NH₃)₄ATP was observed (Fig. 3). The calculated dissociation constant in this case (low affinity site), however, was by about one order of magnitude higher (K_d=0.2 mM).

The extinction coefficient of FEDA-ATP in 20 mM Tris/HCl buffer (pH 9) was $A_{490}=78,000 \text{ M}^{-1}$ cm⁻¹. The fluorescence measurements in 20 mM Tris/HCl buffer (pH 7.25) result in an $\lambda_{ex}=495$ nm and $\lambda_{em}=520$ nm. We titrated a solution of 1 μ M FEDA-ATP with increasing concentrations of Na⁺/K⁺-ATPase but were not able to detect any significant change of the steady-state fluorescence intensity even with a 10-fold excess of the enzyme (data not shown).



Fig. 4. Inactivation of Na^+/K^+ -ATPase activity by $Cr(H_2O)_4FEDA$ -ATP. One unit of Na^+/K^+ -ATPase was incubated in a total volume of 1 ml at 37 °C with 20 mM Tris/HCl-buffer pH 7.25, 15 mM NaCl and various concentrations of the inhibitor (full square control, open square 0.3 μ M, full circle 1.3 μ M and open circle 5.0 μ M $Cr(H_2O)_4FEDA$ -ATP). Residual activity was measured by transferring aliquots of 50 μ l to the optical assay mixture (Schoner et al. 1967).

To bring about irreversible binding, we prepared analogs of FEDA-ATP with chromium(III) and cobalt (III) and studied their inactivation effect on Na⁺/K⁺-ATPase. One unit of the enzyme was incubated in 20 mM Tris/HCl (pH 7.25), 15 mM NaCl

at 37 °C with increasing concentrations of Cr(H₂O)₄FEDA-ATP (Fig. 4) and Co(NH₃)₄FEDA-ATP (Fig. 5). Aliquots of 50 μ l were taken from the incubation medium for the residual Na⁺/K⁺-ATPase activity measurement by the optical assay. The Mg²⁺ complex derivatives of FEDA-ATP behaved similarly to those of ATP: 5 μ M Cr(H₂O)₄FEDA-ATP inactivated by 55% in 3 h (Fig. 4), while 6 μ M Co(NH₃)₄FEDA-ATP inactivated by 30% only (Fig. 5).



Fig. 5. Inactivation of Na^+/K^+ -ATPase activity by $Co(NH_3)_4FEDA$ -ATP. One unit of Na^+/K^+ -ATPase was incubated in a total volume of 1 ml at 37 °C with 20 mM Tris/HCl-buffer pH 7.25, 15 mM NaCl and various concentrations of inhibitor (full square control, cross 3.0 μ M, full rhombus 6.0 μ M and open rhombus 12.5 μ M $Co(NH_3)_4FEDA$ -ATP). Residual activity was measured by transferring aliquots of 50 μ l to the optical assay mixture (Schoner et al. 1967).

Discussion

FEDA-ATP does not appear to be a substrate of Na⁺/K⁺-ATPase isolated from the pig kidney but rather a reversible competitive inhibitor of ATP hydrolysis. The presence of FEDA-ATP resulted in enzyme inhibition but the steady-state fluorescence intensity was not changed significantly. This is in contrast with previous reports on myosin fragment S1 (Conibar *et al.* 1996) where FEDA-ATP was reported as a substrate with changed fluorescence properties after binding. This could indicate differences in the microenvironments of the binding sites of the two systems. Nevertheless, additional steady-state and dynamic fluorescence studies will have to be performed.

On the other hand, FEDA-ATP showed the affinity for both the low (E₂, K_d=200 μ M) and the high (E₁, K_d=2 μ M) affinity ATP-binding sites. Apparently, the FEDA-ATP behaves like ATP in these reactions (Schoner *et al.* 1994). When the microscopic affinities of FEDA-ATP were used to calculate the macroscopic

affinity in the overall reaction according to $K_i = (K_{dE1} * K_{dE2})^{1/2}$ given by Thoenges *et al.* (1997), the experimentally measured inhibition constant of 66 μ M was obtained.

In addition, the chromium(III) and cobalt(III) analogs of FEDA-ATP behaved similarly as the analogs of ATP. The chromium analog of FEDA-ATP inhibited the high affinity ATP-binding site while the cobalt analog of FEDA-ATP was found to inhibit the low-affinity ATP-binding site. These observations open up the possibility for labelling specifically the E₁ATP site with $Cr(H_2O)_4$ FEDA-ATP and to introduce another label on the E₂ATP site as a Co(NH₃)₄ATP derivative. Since 2'(3')-O-[amino ethyl carbamoyl] adenosine 5'-triphosphate (EDA-ATP) can react with any other isothiocyanate but FITC (Cremo *et al.* 1990), other fluorescent ATP analogs overlapping in their fluorescent spectra with fluorescein are conceivable as $Co(NH_3)_{4}$ complex derivatives. Such analogs may help to determine the distance between the E_1ATP and E_2ATP sites by Förster energy transfer.

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